

Applications of 4,4'-(Me₃Si)₂-BINAP in Transition-Metal-Catalyzed Asymmetric Carbon–Carbon Bond-Forming Reactions

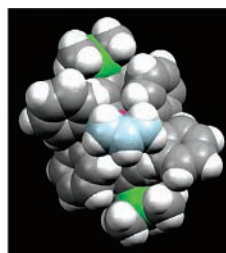
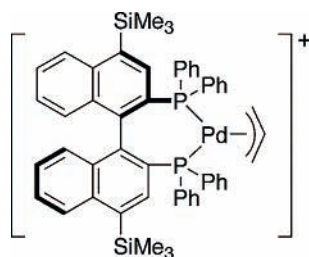
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ABSTRACT



A recently developed BINAP derivative with trimethylsilyl substituents on the 4- and 4'-positions of the binaphthyl skeleton, 2,2'-bis(diphenylphosphino)-4,4'-bis(trimethylsilyl)-1,1'-binaphthyl (tms-BINAP), was used in a variety of transition-metal-catalyzed asymmetric carbon–carbon bond-forming reactions. In π -allylpalladium-mediated reactions, tms-BINAP gave better enantioselectivity than the unsubstituted BINAP, and the origin of the improved enantioselectivity was gained from an X-ray structural study of [Pd(η^3 -C₃H₅)]((R)-tms-BINAP)[ClO₄].

Design of chiral ligands is central to the development of transition-metal-catalyzed asymmetric reactions. Among numerous reported chiral ligands, 2,2'-bis(diphenylphosphino)-1,1'-binaphthyl (BINAP) is arguably the most successful one to date.¹ Since the first report of BINAP on the rhodium-catalyzed asymmetric hydrogenation in 1980,^{1a} transition-metal/BINAP complexes have been used for a wide range of asymmetric reactions with good enantioselectivity, which include Rh-catalyzed isomerization of allylamines,² Ru-catalyzed hydrogenation of carbonyl groups,³ Pd-catalyzed Heck reaction,⁴ Rh-catalyzed conjugate addition

of aryl- or alkenyl-nucleophiles,⁵ Ir-catalyzed Pauson–Khand type cyclization,⁶ and Ag-catalyzed allylation of carbonyl

(2) (a) Tani, K.; Yamagata, T.; Otsuka, S.; Akutagawa, S.; Kumobayashi, H.; Taketomi, T.; Takaya, H.; Miyashita, A.; Noyori, R. *J. Chem. Soc., Chem. Commun.* **1982**, 600. (b) Akutagawa, S. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999; Vol. 2, Chapter 23.

(3) (a) Ohkuma, T.; Ooka, H.; Hashiguchi, S.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1995**, 117, 2675. (b) Ohkuma, T.; Noyori, R. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999; Vol. 1, Chapter 6.1.

(4) (a) Sato, Y.; Sodeoka, M.; Shibasaki, M. *J. Org. Chem.* **1989**, 54, 4738. (b) Carpenter, N. E.; Kucera, D. J.; Overman, L. E. *J. Org. Chem.* **1989**, 54, 5846. (c) Ozawa, F.; Kubo, A.; Hayashi, T. *J. Am. Chem. Soc.* **1991**, 113, 1417. (d) Shibasaki, M.; Vogl, E. M. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999; Vol. 1, Chapter 14.

(5) (a) Takaya, Y.; Ogasawara, M.; Hayashi, T.; Sakai, M.; Miyauchi, N. *J. Am. Chem. Soc.* **1998**, 120, 5579. (b) Hayashi, T.; Yamasaki, K. *Chem. Rev.* **2003**, 103, 2829. (c) Hayashi, T. *Bull. Chem. Soc. Jpn.* **2004**, 77, 13.

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(1) (a) Miyashita, A.; Yasuda, A.; Takaya, H.; Toriumi, K.; Ito, T.; Souchi, T.; Noyori, R. *J. Am. Chem. Soc.* **1980**, 102, 7932. (b) Noyori, R.; Takaya, H. *Acc. Chem. Res.* **1990**, 23, 345.

groups.⁷ Structural modification of BINAP has been extensively examined to improve enantioselectivity of the asymmetric reactions catalyzed by transition-metal/BINAP complexes. Notable examples include modifications of the phenyl groups of the PPh_2 moieties of BINAP⁸ and the development of atropisomeric bisphosphines based on modified biaryls such as $\text{H}_8\text{-BINAP}$,⁹ MeO-biphep ,¹⁰ biphemp ,¹¹ and seghpos .¹²

Recently, we have reported a novel strategy of BINAP modification by introducing sterically encumbered substituents at the 4- and 4'-positions of the binaphthyl skeleton, which drastically enhances enantioselectivity in the Ru-catalyzed asymmetric hydrogenation of a variety of carbonyl compounds.¹³ Herein, we wish to report the effectiveness of this novel class of modified BINAPs in transition-metal-catalyzed asymmetric carbon-carbon bond-forming reactions. Four different asymmetric carbon-carbon bond-forming reactions were examined in this work: (1) Pd-catalyzed asymmetric synthesis of axially chiral allenes from 2-bromo-1,3-dienes,¹⁴ (2) Pd-catalyzed asymmetric allylation of prochiral nucleophiles,¹⁵ (3) Rh-catalyzed conjugate addition of ArB(OH)_2 to α,β -unsaturated carboxylic esters,^{5b-c,16} and (4) Pd-catalyzed asymmetric allylic alkylation reactions.¹⁷ In the first three reactions, BINAP has shown superiority over other chiral phosphines; however, the reported enantioselectivity still has room for further improvement.

Among the 4,4'-disubstituted BINAPs, those with Me_3Si - or $(\text{HO})_2\text{P(O)}$ -substituents have been shown to give the highest enantioselectivity in Ru-catalyzed asymmetric hydrogenation reactions.¹³ Since some of the reactions examined here require aprotic conditions, 4,4'-(Me_3Si)₂-BINAP (tms-BINAP) was chosen as a representative of the 4,4'-disubstituted BINAPs for the present study (Figure 1).

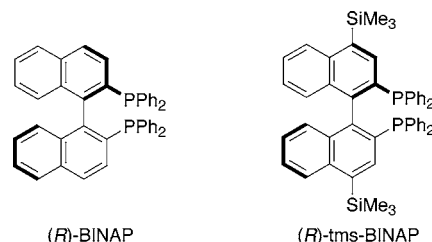


Figure 1. BINAP and 4,4'-(Me_3Si)₂-BINAP (tms-BINAP).

The effect of tms -substituents was first examined in the Pd-catalyzed asymmetric synthesis of axially chiral allenes.¹⁴ For a direct comparison between BINAP and tms-BINAP , two reactions, one with BINAP and the other with tms-BINAP , were set up simultaneously and carried out side by side under identical conditions. The results are summarized in Table 1. In the previous report on the asymmetric allene

- (6) Shibata, T.; Takagi, K. *J. Am. Chem. Soc.* **2000**, *122*, 9852.
 (7) Yanagisawa, A.; Nakashima, H.; Ishiba, A.; Yamamoto, H. *J. Am. Chem. Soc.* **1996**, *118*, 4723.
 (8) (a) Takaya, H.; Mashima, K.; Koyano, K.; Yagi, M.; Kumobayashi, H.; Taketomi, T.; Akutagawa, S.; Noyori, R. *J. Org. Chem.* **1986**, *51*, 629. (b) Mashima, K.; Kusano, K.; Sato, N.; Matsumura, Y.; Nozaki, K.; Kumobayashi, H.; Sayo, N.; Hori, T.; Ishizaki, T.; Akutagawa, S.; Takaya, H. *J. Org. Chem.* **1994**, *59*, 3064. (c) Gladiali, S.; Dore, A.; Fabbri, D.; Medici, S.; Pirri, G.; Pulacchini, S. *Eur. J. Org. Chem.* **2000**, 2861.
 (9) (a) Zhang, X.; Mashima, K.; Koyano, K.; Sayo, N.; Kumobayashi, H.; Akutagawa, S.; Takaya, H. *Tetrahedron Lett.* **1991**, *32*, 7283. (b) Zhang, X.; Mashima, K.; Koyano, K.; Sayo, N.; Kumobayashi, H.; Akutagawa, S.; Takaya, H. *J. Chem. Soc., Perkin Trans. 1* **1994**, 2309.
 (10) Schmid, R.; Foricher, J.; Cereghetti, M.; Schönholzer, P. *Helv. Chim. Acta* **1991**, *74*, 370.
 (11) (a) Schmid, R.; Cereghetti, M.; Heiser, B.; Schönholzer, P.; Hansen, H. *J. Helv. Chim. Acta* **1988**, *71*, 897. (b) Frejd, T.; Klingstedt, T. *Acta Chem. Scand.* **1989**, *43*, 670.
 (12) Saito, T.; Yokozawa, T.; Ishizaki, T.; Moroi, T.; Sayo, N.; Miura, T.; Kumobayashi, H. *Adv. Synth. Catal.* **2001**, *343*, 264.
 (13) (a) Hu, A.; Ngo, H. L.; Lin, W. *Angew. Chem., Int. Ed.* **2004**, *43*, 2501. (b) Hu, A.; Ngo, H. L.; Lin, W. *Org. Lett.* **2004**, *6*, 2937. (c) Hu, A.; Lin, W. *Org. Lett.* **2005**, *7*, 455. (d) Ngo, H. L.; Lin, W. *J. Org. Chem.* **2005**, *70*, 1177.
 (14) (a) Ogasawara, M.; Ikeda, H.; Hayashi, T. *Angew. Chem., Int. Ed.* **2000**, *39*, 1042. (b) Ogasawara, M.; Ikeda, H.; Nagano, T.; Hayashi, T. *Org. Lett.* **2001**, *3*, 2615. (c) Ogasawara, M.; Ikeda, H.; Nagano, T.; Hayashi, T. *J. Am. Chem. Soc.* **2001**, *123*, 2089. (d) Ogasawara, M.; Ueyama, K.; Nagano, T.; Mizuhata, T.; Hayashi, T. *Org. Lett.* **2003**, *5*, 217.
 (15) (a) Kuwano, R.; Ito, Y. *J. Am. Chem. Soc.* **1999**, *121*, 3236. (b) Kuwano, R.; Nishio, R.; Ito, Y. *Org. Lett.* **1999**, *1*, 837. (c) Kuwano, R.; Uchida, K.; Ito, Y. *Org. Lett.* **2003**, *5*, 2177.
 (16) Takaya, Y.; Senda, T.; Kurushima, H.; Ogasawara, M.; Hayashi, T. *Tetrahedron: Asymmetry* **1999**, *10*, 4047.
 (17) For reviews: (a) Trost, B. M.; Chulbom, L. In *Catalytic Asymmetric Synthesis*, 2nd ed.; Ojima, I., Ed.; VCH: New York, 2000; p 593. (b) Pfaltz, A.; Lautens, M. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999; Vol. 2, Chapter 24. (c) Trost, B. M.; Van Vranken, D. L. *Chem. Rev.* **1996**, *96*, 395. (d) Hayashi, T. In *Catalytic Asymmetric Synthesis*; Ojima, I., Ed.; VCH: New York, 1993; p 325. (e) Frost, C. G.; Howarth, J.; Williams, J. M. J. *Tetrahedron: Asymmetry* **1992**, *3*, 1089.

Table 1. Pd-Catalyzed Asymmetric Synthesis of Axially Chiral Allenes^a

entry	1	2	base	L*	solvent	yield ^b /%	% ee ^c (config) ^d
1a	1a	2m	CsO ^t Bu	(R)-BINAP	CH ₂ Cl ₂	70 (3am)	74 (R)
1b				(R)-tms-BINAP		72 (3am)	85 (R)
2a	1a	2n	NaH	(R)-BINAP	THF	80 (3an)	70 (R)
2b				(R)-tms-BINAP		82 (3an)	80 (R)
3a ^{e,f}	1b	2o	CsO ^t Bu	(R)-BINAP	THF	73 (3bo)	53 (R)
3b ^{e,f}				(R)-tms-BINAP		83 (3bo)	61 (R)
4a	1c	2p	KH	(R)-BINAP	THF	76 (3cp)	62 (R)
4b				(R)-tms-BINAP		98 (3cp)	77 (R)

^a All the reaction were carried out with **1** (0.50 mmol), **2** (0.55 mmol), and base (0.55 mmol) in a given solvent (5.0 mL) for 24 h in the presence of a Pd catalyst (10 mol %) generated from $\text{Pd}(\text{dba})_2$ and the chiral phosphine. ^b Isolated yield by chromatography on alumina. ^c Determined by chiral HPLC (Chiralpak AD-H (**3am** and **3an**), Chiralcel OD-H (**3bo** and **3cp**)). ^d The absolute configurations were deduced by the Lowe-Brewster rule (ref 19). ^e With 5 equiv of **2o** with respect to **1b**. ^f At 0 °C.

synthesis, malonate derivatives, such as **2m** and **2n**, were used as pronucleophiles.^{14c,d} While the Pd/BINAP catalyst gave the axially chiral allene **3am** with 74% ee for the reaction of the ^tBu-substituted bromodiene **1a** with **2m** (entry

1a), the Pd/tms-BINAP catalyst afforded **3am** in 85% ee under the identical conditions except the ligand (entry 1b). Similarly, tms-BINAP gave higher ee's than BINAP for the reaction of **1a** and **2n** to give **3an** (80% ee vs 70% ee; entries 2a and 2b) and **1b** and **2o** to give **3bo** (61% ee vs 53% ee; entries 3a and 3b). An N-nucleophile generated from HN-(boc)₂ (**2o**) and KH was found to be effective for the asymmetric reaction.¹⁸ The asymmetric amination product **3cp** of 62% ee was obtained in 76% yield using BINAP (entry 4a). In comparison, the Pd/tms-BINAP system afforded **3cp** with 77% ee in 98% yield under the same conditions (entry 4b).

Encouraged by these results, we have examined the effects of tms-BINAP on Pd-catalyzed allylations of prochiral nucleophiles.¹⁵ BINAP was reported to be a particularly effective chiral ligand for the reactions using α -acetamido- β -ketoesters as pronucleophiles. As shown in Table 2, the

Table 2. Pd-Catalyzed Asymmetric Allylation of Prochiral Nucleophiles^a

$\text{R} = \text{Me (4a)}$ $\text{R}' = \text{H (5m)}$
 $\text{R} = \text{Ph (4b)}$ $\text{R}' = \text{Ph (5n)}$

entry	4	5	L*	yield ^b /%	% ee ^c (config) ^d
1a	4a	5m	(R)-BINAP	87 (6am)	68 (R)
1b	4a	5m	(R)-tms-BINAP	75 (6am)	77 (R)
2a	4a	5n	(R)-BINAP	78 (6an)	90 (R)
2b	4a	5n	(R)-tms-BINAP	68 (6an)	93 (R)
3a	4b	5m	(R)-BINAP	93 (6bm)	72 (R)
3b	4b	5m	(R)-tms-BINAP	90 (6bm)	84 (R)

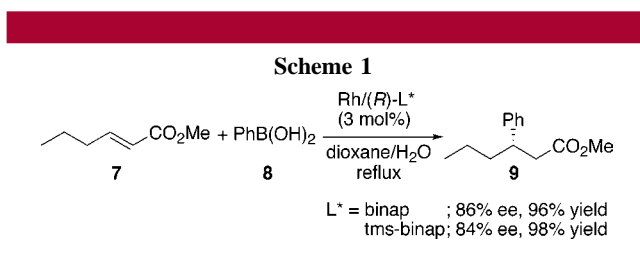
^a All the reactions were carried out with **4** (0.50 mmol) and **5** (0.80 mmol) at $-25\text{ }^{\circ}\text{C}$ ²⁰ in toluene (2.5 mL) in the presence of a Pd catalyst (1 mol %) generated from [PdCl(π -allyl)]₂ and the chiral phosphine. ^b Isolated yield by silica gel chromatography. ^c Determined by chiral HPLC (Chiralcel OD-H (**6am** and **6an**), Chiralcel OJ-H (**6bm**)). ^d Determined on the basis of the sign of the specific rotations of the products.

Pd-catalyst with tms-BINAP gave higher ee's for these reactions. For instance, the allylation product **6am** of 68% ee was obtained by a Pd/BINAP-catalyzed reaction of **4a** and **5m** in toluene at $-25\text{ }^{\circ}\text{C}$ (entry 1a).²⁰ In comparison, the Pd/tms-BINAP catalyst afforded **6am** of 77% ee under the same conditions (entry 1b). Analogously, the Pd/tms-BINAP catalyst exhibited higher ee's than the Pd/BINAP catalyst for the reactions of **4a** and **5n**, giving **6an** (93% ee

vs 90% ee; entries 2a and 2b), and **4b** and **5m**, giving **6bm** (84% ee vs 72% ee; entries 3a and 3b).

Contrary to the above-mentioned Pd-catalyzed reactions, tms-BINAP did not give ee enhancement for the Rh-catalyzed conjugate addition of arylboronic acids to α,β -unsaturated carboxylic esters.¹⁶ A representative example was shown in Scheme 1. For the reaction of methyl 2-hexenoate (**7**) with PhB(OH)₂ (**8**), the Rh/BINAP catalyst gave methyl 3-phenylhexanoate (**9**) of 86% ee in 96% yield. The enantioselectivity of the Rh/tms-BINAP was slightly lower for the same reaction, and the addition product **9** was obtained in 98% yield with 84% ee.

It is known that the key intermediates for the two Pd-catalyzed reactions in which tms-BINAP are effective are quite similar to each other. The reactions in Table 2 proceeded via a well-known π -allylpalladium intermediate.^{15a} The intermediate for the Pd-catalyzed allene formation process (Table 1) is a (1,2,3- η^3 -butadien-3-yl)palladium species,^{14a,c} which possesses a π -allylpalladium substructure as well. Apparently, the effect of the Me₃Si substituents was not operative in the Rh-catalyzed reaction (Scheme 1), of



which the catalytic cycle²¹ as well as the structure of the suggested stereodetermining intermediates^{5a,16} were different from the above-mentioned π -allylpalladium-mediated reactions.

To gain insight into the origin of the interesting enantio-enhancement of the tms-BINAP ligand in the π -allylpalladium-mediated asymmetric reactions, we have prepared [Pd(η^3 -C₃H₅)((R)-tms-BINAP)]ClO₄ and determined its structure by single-crystal X-ray diffraction studies.²² As shown in Figure 2 for the space-filling model of the [Pd(η^3 -C₃H₅)-((R)-tms-BINAP)]⁺ ion, there is steric interaction between the SiMe₃ substituents on the binaphthyl skeleton and the phenyl groups of the diphenylphosphino moieties. Consequently, the average dihedral angle of 76.7° between the naphthyl rings in [Pd(η^3 -C₃H₅)((R)-tms-BINAP)](ClO₄) is smaller than that between the naphthyl rings in the BINAP analogue (79.9°).^{15a} This change in the dihedral angle can in effect tilt the equatorial phenyl groups toward the coordinating η^3 -allyl moiety to presumably lead to a better stereodiscrimination between the favored and disfavored

(21) Hayashi, T.; Takahashi, M.; Takaya, Y.; Ogasawara, M. *J. Am. Chem. Soc.* **2002**, *124*, 5052.

(22) η^3 -C₃H₅[(R)-tms-BINAP]ClO₄ was prepared according to the literature procedure for its BINAP analogue and crystallizes (from slow evaporation of a CH₂Cl₂/2-propanol solution) in the chiral space group P2₁ with two molecules in each asymmetric unit. See Pregosin, P. S.; Ruegger, H.; Salzmann, R.; Albinati, A.; Lianza, F.; Kunz, R. W. *Organometallics* **1994**, *13*, 83.

(18) Recently, Imada et al. reported dynamic kinetic resolution of racemic allenylmethyl esters by the Pd-catalyzed amination giving allenic amines; see Nishida, M.; Kutsuwa, K.; Imada, Y.; Murahashi, S.-I. Naota, T. In *Abstracts*; 51st Symposium on Organometallic Chemistry, October 2–3, 2004, Tokyo, Japan; Kinki Chemical Society: Japan, 2004; PB153. Also see Imada, Y.; Ueno, K.; Kutsuwa, K.; Murahashi, S.-I. *Chem. Lett.* **2002**, 140.

(19) (a) Lowe, G. *Chem. Commun.* **1965**, 411. (b) Brewster, J. H. *Top. Stereochem.* **1967**, *2*, 1.

(20) In the original report by Kuwano and Ito,^{15a} the reactions were performed at $-30\text{ }^{\circ}\text{C}$. Because of a limitation of the equipment available in our laboratory, we carried out the reactions at $-25\text{ }^{\circ}\text{C}$.

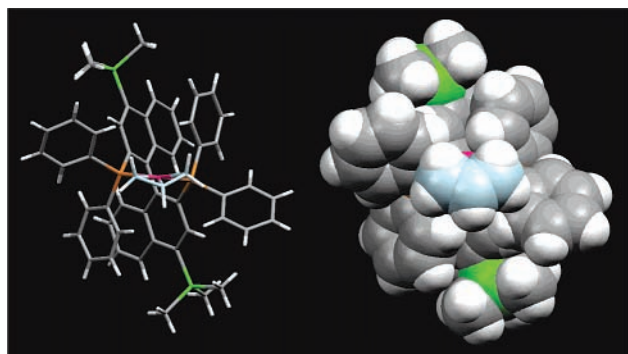


Figure 2. Capped stick and space-filling models of single-crystal X-ray structure of $[\text{Pd}(\eta^3\text{-allyl})((R)\text{-tms-BINAP})]\text{ClO}_4$. Red, Ru; orange, P; green, Si; gray, C; white, H. The allyl carbon atoms are highlighted in light blue color. The ClO_4^- anion is omitted for clarity.

diastereomeric transition states during the subsequent step of nucleophilic attack on the π -allyl group.

The X-ray structure study as well as the results in Tables 1 and 2 prompted us to explore the effect of tms-BINAP on Pd-catalyzed asymmetric allylic alkylations.¹⁷ The allylic alkylation reactions shown in Table 3 proceed via an essentially identical intermediate to that of the reactions in Table 2, with the key difference of generating a stereogenic center on the electrophile fragment. Previous studies indicated that BINAP was not a suitable ligand for this asymmetric reaction, and the Pd/BINAP catalyst showed only modest enantioselectivity for most cases.²³ As shown in Table 3, while the Pd/BINAP catalyst gave the alkylated product **12am** with only 25% ee for the reaction of 1,3-diphenyl-2-propenyl acetate (**10a**) with sodium dimethyl methylmalonate (**11m**) (entry 1a), the Pd/tms-BINAP catalyst afforded **12am** of a much higher 80% ee under the same conditions (entry 1b). Similarly, tms-BINAP showed a better enantioselectivity than BINAP for reactions of **10a** and **11n**, giving **12an** (94% ee vs 84% ee; entries 2a and 2b), and **10b** and **11m**, giving **12bm** (57% ee vs 40% ee; entries 3a and 3b).

In summary, we have applied 4,4'-(Me_3Si)₂-BINAP in several Pd- and Rh-catalyzed asymmetric carbon–carbon

(23) (a) Yamaguchi, M.; Shima, T.; Yamagishi, T.; Hida, M. *Tetrahedron Lett.* **1990**, *31*, 5049. (b) Yamaguchi, M.; Shima, T.; Yamagishi, T.; Hida, M. *Tetrahedron: Asymmetry* **1991**, *2*, 663.

Table 3. Pd-Catalyzed Asymmetric Allylic Alkylations^a

$\text{Ph}-\text{CH}=\text{CH}-\text{Ph} \xrightarrow[\text{THF}]{\text{Pd}/(R)\text{-L}^* (2 \text{ mol}\%)} \text{Ph}-\text{CH}(\text{Nu})=\text{CH}-\text{Ph}$						
	10a	11m-n				12am-an
$\text{Cyclohexene-OAc} \xrightarrow[\text{THF}]{\text{Pd}/(R)\text{-L}^* (2 \text{ mol}\%)} \text{Cyclohexene-Nu}$						
	10b	11m				12bm
Nu-M^+ : $\text{Na}[\text{CMe}(\text{COOMe})_2]$ (11m) $\text{Cs}[\text{C}(\text{NHAc})(\text{COOEt})_2]$ (11n)						
entry	10	11	L*	conditions	yield ^b /%	% ee ^c (config) ^d
1a	10a	11m	(<i>R</i>)-BINAP	rt, 12 h	85 (12am)	25 (<i>R</i>)
1b			(<i>R</i>)-tms-BINAP		96 (12am)	80 (<i>R</i>)
2a	10a	11n	(<i>R</i>)-BINAP	0 °C, 24 h	89 (12an)	84 (<i>R</i>)
2b			(<i>R</i>)-tms-BINAP		>99 (12an)	94 (<i>R</i>)
3a	10b	11m	(<i>R</i>)-BINAP	0 °C, 24 h	91 (12bm)	40 (<i>S</i>)
3b			(<i>R</i>)-tms-BINAP		94 (12bm)	57 (<i>S</i>)

^a All the reactions were carried out with **10** (0.50 mmol) and **11** (0.55 mmol) in THF (5.0 mL) in the presence of a palladium catalyst (2 mol %) generated from $[\text{PdCl}(\pi\text{-allyl})_2]$ and the chiral phosphine. ^b Isolated yield by silica gel chromatography. ^c Determined by chiral HPLC on a Chiralpak AD-H column. ^d Determined on the basis of the sign of the specific rotations of the products.

bond-forming reactions. It was found that tms-BINAP was more enantioselective than the unsubstituted BINAP in the π -allylpalladium-mediated reactions; however, the effect of the Me_3Si -substituents was not operative in the Rh-catalyzed conjugate addition of phenylboronic acid to an α,β -unsaturated ester. A comprehensive survey of the scope and the limitation of the 4,4'-substituted-BINAP in other transition-metal-catalyzed asymmetric transformations is currently under investigation.

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Supporting Information Available: Detailed experimental procedures, compound characterization data, and crystallographic data (CIF file). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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